



# Datasheet Quinoxaline-2-carboxylic acid-d<sub>4</sub>

**Reference number: CEC/MAT: 23** 

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source : CSL

"Bank of Reference Standards"

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 $d_0 < 1 \%$ 

Figure 1. Molecular structure of quinoxaline-2-carboxylic acid-d<sub>4</sub>

Name : quinoxaline-[5,6,7,8-d<sub>4</sub>]-2-carboxylic acid Synonym : (benzene ring-d<sub>4</sub>)-quinoxaline-2-carboxylic acid

 $\begin{array}{ll} \mbox{Molecular formula} & : C_9 H_2 D_4 N_2 O_2 \\ \mbox{Molecular weight} & : 178.183 \end{array}$ 

Long term stability tested on 1997.10.08 :  $91.3 \pm 5.6 \%$  (storage 4 °C, analysis HPLC-UV, 6 test on 2 ampoules)

#### Methods of Characterization:

I UV spectroscopy
 II IR spectroscopy
 III Mass spectroscopy
 IV 'H-NMR spectroscopy

## I UV Spectroscopy

Instrument: Hitachi U 3000

Method: Dissolved in ethanol (10mg/l)

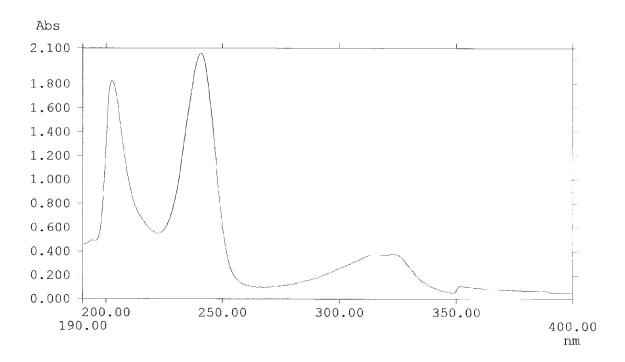


Figure 2. The UV spectrum of quinoxaline-2-carboxylic acid-d<sub>4</sub>

Peak no.	wavelength (nm)	absorbance
1	352.00	0.1123
2	323.00	0.3751
3	240.50	2.0549
4	202.50	1.8279

## II IR-Spectroscopy

Instrument: Perkin Elmer STIR 1720X

Sampling technique: nujol mull

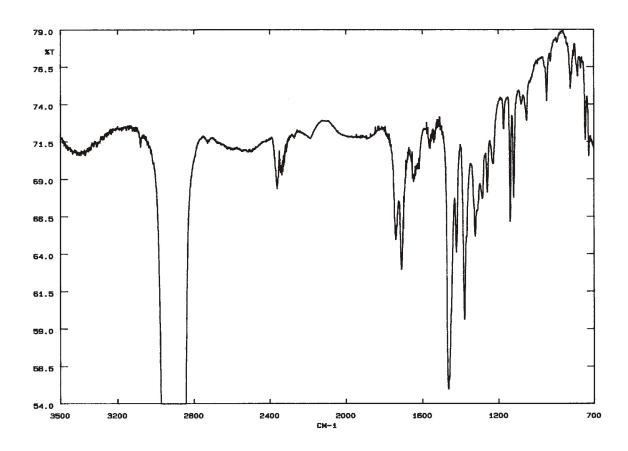


Figure 3. The IR-spectrum of quinoxaline-2-carboxylic acid-d<sub>4</sub>

Wavelength (cm-1)	designation	
3369	O-H stretch	
1739		
1710	C=O stretch	
1421		
1322		
1281		
1259		
1230		
1139		
1120		
947		
745		
727		

## III Mass Spectroscopy

Instrument: Kratos MS 25

Sampling technique: Direct probe, 70 ev electron impact

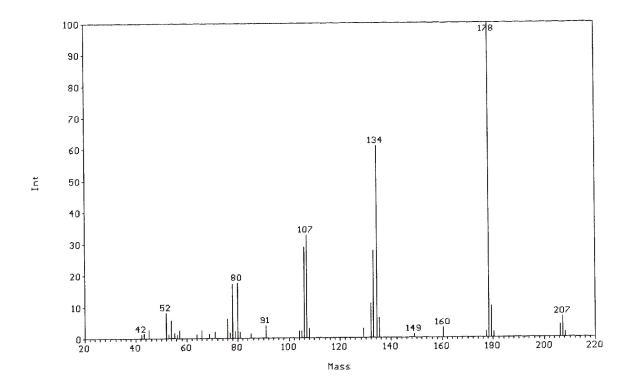


Figure 4. The mass spectrum of quinoxaline-2-carboxylic acid-d<sub>4</sub>

m/u	percentage	designation
178	100M	
134	61	M - CO <sub>2</sub>
133	28	M - CO <sub>2</sub> H
107	33	$M - (CO_2 + HCN?)$
106	29	$M - (CO_2H + HCN?)$
80	18	$C_6D_4$ or $C_4N_2H_4$ ?
77	17	$C_6H_5$ ?

Datasheet BCR-MAT reference standard material Quinoxaline-2-carboxylic acid-d<sub>4</sub>

## III <sup>1</sup>H-NMR Spectroscopy

Instrument: GX 400

Solvent: acetone- $d_6$  +  $CD_3OD$  with TMS (d = 0.0) as internal standard.

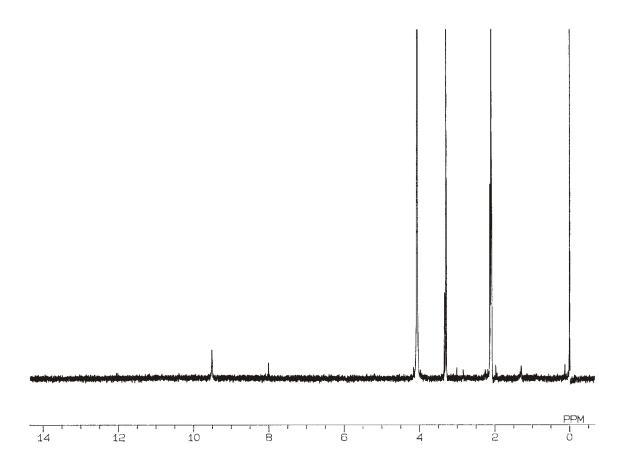


Figure 5. The NMR spectrum of quinoxaline-2-carboxylic acid-d<sub>4</sub>

Chemical Shift (d)	number of protons	designation
9.51	1	H3 (arom.)

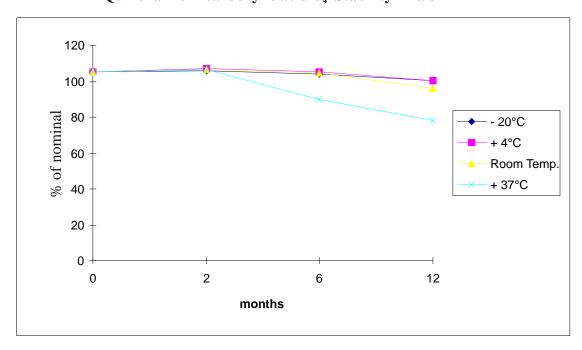
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#### Preparation and validation of reference standards

The production of ampoules containing the reference material was described in the final report produced September 1995.

The thermal stability of the compound was under investigation and the results for quinoxaline-2-carboxylic acid-d<sub>4</sub> over a period of 12 months are expressed in the table and chart below.

#### Quinoxaline-2-carboxylic acid-d<sub>4</sub> Stability Trials



#### Results of quinoxaline-2-carboxylic acid-d<sub>4</sub> stability trials

The results below are the % recovery (with cv) of quinoxaline-2-carboxylic acid-d<sub>4</sub> at 4 different storage temperatures over a period of 12 months compared with a standard equivalent to 0.1 mg.

	temp. (°C)			t = 6 months (% recovery)	t = 12 months (% recovery)
QCA-d <sub>4</sub>	- 20°C	105.5 +/- 1.5	106.0 +/- 2.0	104.0 +/- 1.0	100.5 +/- 1.0
	4°C	-	107.0 +/- 1.5	105.5 +/- 1.5	100.5 +/- 1.5
	Room Temp.	-	106.5 +/- 2.0	104.5 +/- 0.5	96.0 +/- 1.0
	37°C	-	106.5 +/- 1.5	90.0 +/- 1.0	78.0 +/- 2.0

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#### Conclusion

The spectroscopic data is consistant with the proposed structure for all the methods of determination although a very small amount of DMF (which was used as solvent during ampouling) was detected by NMR at approximately 2.8 and 2.9 d. There was also a group of 3 unexpected peaks at 206, 207 and 208 m/u (1.5, 6.5 and 4 % resp.) in the MS spectrum which were probably due to some form of bonding or association with the DMF (Note - M + NCH $_3$  = 207 m/u).

No significant impurities were detected by any of the methods of characterization employed.

The results from the stability trials indicate that quinoxaline-2-carboxylic acid- $d_4$  is acceptably stable over a period of 12 months except for the ampoules stored at + 37 °C which had lost an average of 15.5 % after 6 months and 28.5 % after 12 months of quinoxaline-2-carboxylic acid- $d_4$  from each ampoule and those stored at room temperature which had lost an average of 9.5 % after 12 months. It is therefore recommended that these ampoules are stored at - 20 °C or lower before opening and at + 4 °C or lower when in solution. The stability of this compound also requires careful monitoring if it is to be used with confidence as a reference standard.